



National
Qualifications
2018

X757/76/02

Physics
Section 1 — Questions

TUESDAY, 8 MAY

9:00 AM – 11:30 AM

Instructions for the completion of Section 1 are given on *page 02* of your question and answer booklet X757/76/01.

Record your answers on the answer grid on *page 03* of your question and answer booklet.

Reference may be made to the Data Sheet on *page 02* of this booklet and to the Relationships Sheet X757/76/11.

Before leaving the examination room you must give your question and answer booklet to the Invigilator; if you do not, you may lose all the marks for this paper.



* X 7 5 7 7 6 0 2 *

DATA SHEET

COMMON PHYSICAL QUANTITIES

Quantity	Symbol	Value	Quantity	Symbol	Value
Speed of light in vacuum	c	$3.00 \times 10^8 \text{ m s}^{-1}$	Planck's constant	h	$6.63 \times 10^{-34} \text{ J s}$
Magnitude of the charge on an electron	e	$1.60 \times 10^{-19} \text{ C}$	Mass of electron	m_e	$9.11 \times 10^{-31} \text{ kg}$
Universal Constant of Gravitation	G	$6.67 \times 10^{-11} \text{ m}^3 \text{ kg}^{-1} \text{ s}^{-2}$	Mass of neutron	m_n	$1.675 \times 10^{-27} \text{ kg}$
Gravitational acceleration on Earth	g	9.8 m s^{-2}	Mass of proton	m_p	$1.673 \times 10^{-27} \text{ kg}$
Hubble's constant	H_0	$2.3 \times 10^{-18} \text{ s}^{-1}$			

REFRACTIVE INDICES

The refractive indices refer to sodium light of wavelength 589 nm and to substances at a temperature of 273 K.

Substance	Refractive index	Substance	Refractive index
Diamond	2.42	Water	1.33
Crown glass	1.50	Air	1.00

SPECTRAL LINES

Element	Wavelength/nm	Colour	Element	Wavelength/nm	Colour
Hydrogen	656	Red	Cadmium	644	Red
	486	Blue-green		509	Green
	434	Blue-violet		480	Blue
	410	Violet	Lasers		
	397	Ultraviolet	<i>Element</i>	<i>Wavelength/nm</i>	<i>Colour</i>
	389	Ultraviolet	Carbon dioxide	9550 } 10590 }	Infrared
Sodium	589	Yellow	Helium-neon	633	Red

PROPERTIES OF SELECTED MATERIALS

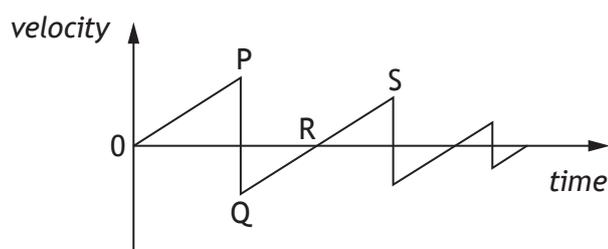
Substance	Density/kg m ⁻³	Melting Point/K	Boiling Point/K
Aluminium	2.70×10^3	933	2623
Copper	8.96×10^3	1357	2853
Ice	9.20×10^2	273
Sea Water	1.02×10^3	264	377
Water	1.00×10^3	273	373
Air	1.29
Hydrogen	9.0×10^{-2}	14	20

The gas densities refer to a temperature of 273 K and a pressure of $1.01 \times 10^5 \text{ Pa}$.

SECTION 1 — 20 marks

Attempt ALL questions

1. A car is moving at a speed of 2.0 m s^{-1} .
The car now accelerates at 4.0 m s^{-2} until it reaches a speed of 14 m s^{-1} .
The distance travelled by the car during this acceleration is
- A 1.5 m
B 18 m
C 24 m
D 25 m
E 48 m.
2. A ball is dropped from rest and allowed to bounce several times.
The graph shows how the velocity of the ball varies with time.



A student makes the following statements about the ball.

- I The ball hits the ground at P.
II The ball is moving upwards between Q and R.
III The ball is moving upwards between R and S.

Which of these statements is/are correct?

- A I only
B II only
C III only
D I and II only
E I and III only

[Turn over

3. A block of mass 6.0 kg and a block of mass 8.0 kg are connected by a string. A force of 32 N is applied to the blocks as shown.



A frictional force of 4.0 N acts on **each** block.

The acceleration of the 6.0 kg block is

- A 1.7 m s^{-2}
B 2.0 m s^{-2}
C 2.3 m s^{-2}
D 2.9 m s^{-2}
E 5.3 m s^{-2} .
4. A person stands on a weighing machine in a lift. When the lift is at rest, the reading on the weighing machine is 700 N.
The lift now descends and its speed increases at a constant rate.
The reading on the weighing machine
- A is a constant value higher than 700 N
B is a constant value lower than 700 N
C continually increases from 700 N
D continually decreases from 700 N
E remains constant at 700 N.
5. Enceladus is a moon of Saturn. The mass of Enceladus is $1.08 \times 10^{20} \text{ kg}$.
The mass of Saturn is $5.68 \times 10^{26} \text{ kg}$.
The gravitational force of attraction between Enceladus and Saturn is $7.24 \times 10^{19} \text{ N}$.
The orbital radius of Enceladus around Saturn is
- A $2.38 \times 10^8 \text{ m}$
B $9.11 \times 10^{13} \text{ m}$
C $5.65 \times 10^{16} \text{ m}$
D $8.30 \times 10^{27} \text{ m}$
E $3.19 \times 10^{33} \text{ m}$.

6. A spacecraft is travelling at $0.10c$ relative to a star.

An observer on the spacecraft measures the speed of light emitted by the star to be

- A $0.90c$
- B $0.99c$
- C $1.00c$
- D $1.01c$
- E $1.10c$.

7. A spacecraft is travelling at a speed of $0.200c$ relative to the Earth.

The spacecraft emits a signal for 20.0 seconds as measured in the frame of reference of the spacecraft.

An observer on Earth measures the duration of the signal as

- A 19.2 s
- B 19.6 s
- C 20.0 s
- D 20.4 s
- E 20.8 s.

8. How many types of quark are there?

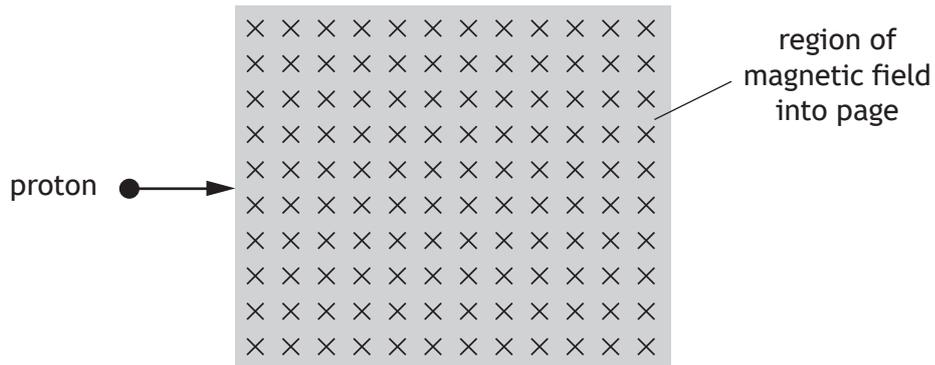
- A 8
- B 6
- C 4
- D 3
- E 2

9. An electron is a

- A boson
- B hadron
- C baryon
- D meson
- E lepton.

[Turn over

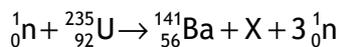
10. A proton enters a region of magnetic field as shown.



On entering the magnetic field the proton

- A deflects into the page
- B deflects out of the page
- C deflects towards the top of the page
- D deflects towards the bottom of the page
- E is not deflected.

11. A nuclear fission reaction is represented by the following statement.



The nucleus represented by X is

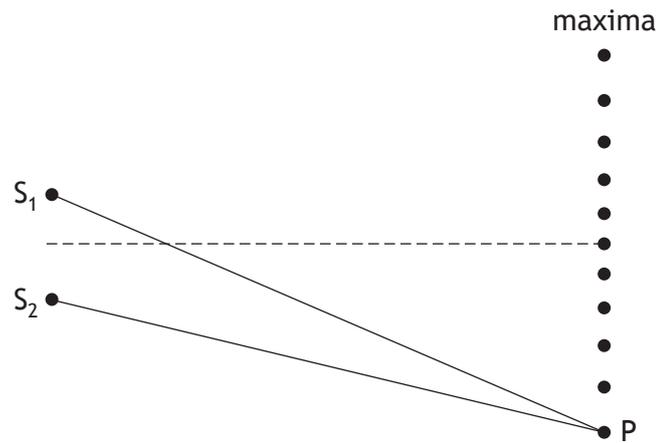
- A ${}_{40}^{96}\text{Zr}$
- B ${}_{36}^{92}\text{Kr}$
- C ${}_{40}^{97}\text{Zr}$
- D ${}_{36}^{93}\text{Kr}$
- E ${}_{40}^{94}\text{Zr}$.

12. The irradiance on a surface 0.50 m from a point source of light is I .

The irradiance on a surface 1.5 m from this source is

- A $0.11I$
- B $0.33I$
- C $1.5I$
- D $3.0I$
- E $9.0I$.

13. Waves from two coherent sources, S_1 and S_2 , produce an interference pattern. Maxima are detected at the positions shown below.



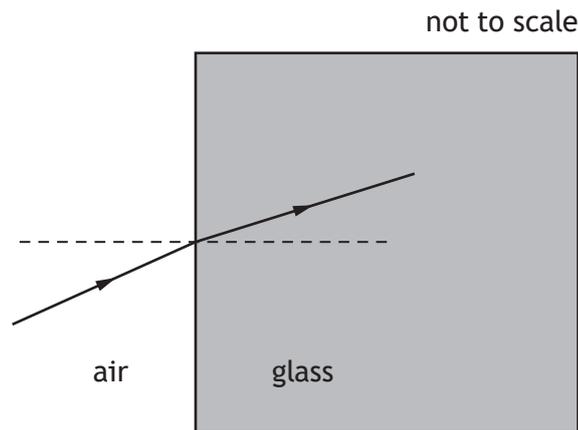
The path difference $S_1P - S_2P$ is 154 mm.

The wavelength of the waves is

- A 15.4 mm
- B 25.7 mm
- C 28.0 mm
- D 30.8 mm
- E 34.2 mm.

[Turn over

14. A ray of monochromatic light passes from air into a block of glass as shown.



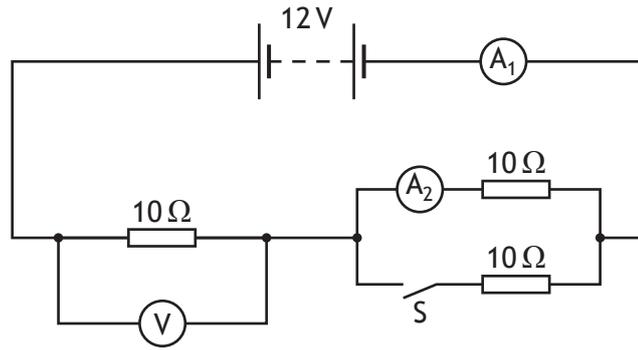
The wavelength of this light in air is 6.30×10^{-7} m.

The refractive index of the glass for this light is 1.50.

The frequency of this light in the glass is

- A 2.10×10^{-15} Hz
- B 1.26×10^2 Hz
- C 1.89×10^2 Hz
- D 4.76×10^{14} Hz
- E 7.14×10^{14} Hz.

15. A circuit is set up as shown.



The battery has negligible internal resistance.

A student makes the following statements about the readings on the meters in this circuit.

- I When switch S is open the reading on the voltmeter will be 6.0 V.
- II When switch S is open the reading on A_2 will be 0.60 A.
- III When switch S is closed the reading on A_1 will be 0.80 A.

Which of these statements is/are correct?

- A I only
- B II only
- C I and II only
- D II and III only
- E I, II and III

16. The power dissipated in a $120\ \Omega$ resistor is 4.8 W.

The current in the resistor is

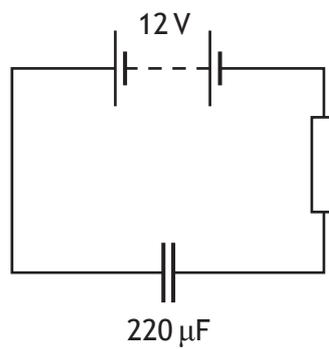
- A 0.020 A
- B 0.040 A
- C 0.20 A
- D 5.0 A
- E 25 A.

[Turn over

17. A $24.0 \mu\text{F}$ capacitor is charged until the potential difference across it is 125 V .
The charge stored on the capacitor is

- A $5.21 \times 10^6 \text{ C}$
- B $7.75 \times 10^{-2} \text{ C}$
- C $1.50 \times 10^{-3} \text{ C}$
- D $3.00 \times 10^{-3} \text{ C}$
- E $1.92 \times 10^{-7} \text{ C}$.

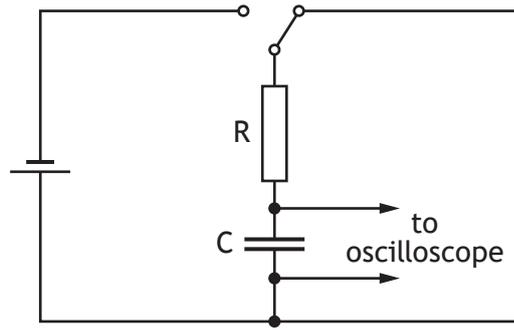
18. A circuit is set up as shown.



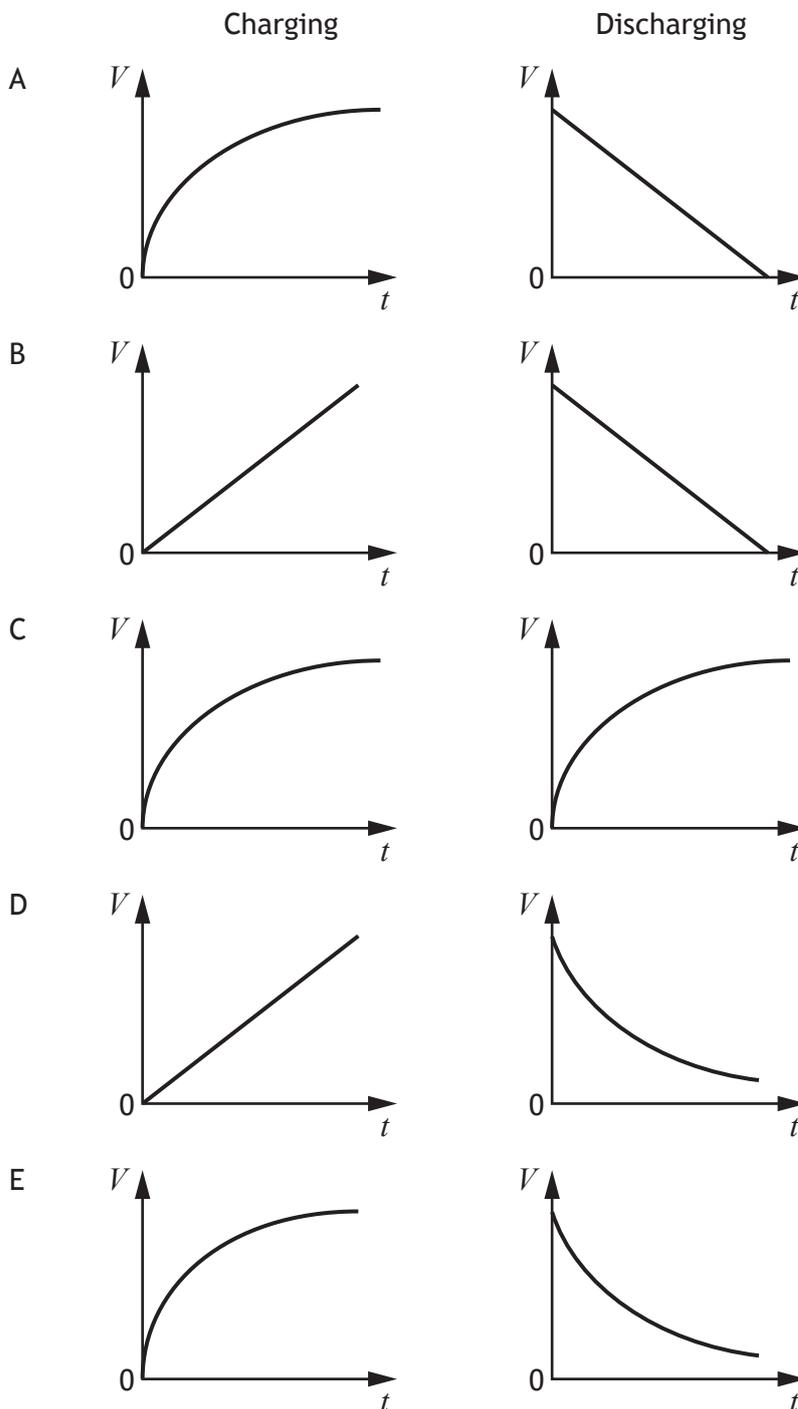
When the capacitor is fully charged the energy stored in the capacitor is

- A $1.6 \times 10^{-5} \text{ J}$
- B $1.3 \times 10^{-3} \text{ J}$
- C $2.6 \times 10^{-3} \text{ J}$
- D $1.6 \times 10^{-2} \text{ J}$
- E $1.6 \times 10^4 \text{ J}$.

19. The circuit shown is used to charge and then discharge a capacitor C.



Which pair of graphs shows how the potential difference V across the capacitor varies with time t during charging and discharging?



20. A student carries out an experiment to determine the specific heat capacity c of a solid. The relationship used to calculate c is

$$c = \frac{E}{m\Delta T}$$

The recorded measurements and their percentage uncertainties are shown.

energy supplied, $E = 5000 \text{ J} \pm 1\%$

mass of solid, $m = 0.20 \text{ kg} \pm 2\%$

change in temperature, $\Delta T = 4.5 \text{ }^\circ\text{C} \pm 5\%$

A good estimate of the percentage uncertainty in the calculated value of c is

- A 8%
- B 7%
- C 5%
- D 3%
- E 1%.

[END OF SECTION 1. NOW ATTEMPT THE QUESTIONS IN SECTION 2
OF YOUR QUESTION AND ANSWER BOOKLET]

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X757/76/01

**Physics
Section 1 — Answer Grid
and Section 2**

TUESDAY, 8 MAY
9:00 AM – 11:30 AM



Fill in these boxes and read what is printed below.

Full name of centre

--

Town

--

Forename(s)

--

Surname

--

Number of seat

--

Date of birth

Day

--	--

Month

--	--

Year

--	--

Scottish candidate number

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Total marks — 130

SECTION 1 — 20 marks

Attempt ALL questions.

Instructions for the completion of Section 1 are given on *page 02*.

SECTION 2 — 110 marks

Attempt ALL questions.

Reference may be made to the Data Sheet on *page 02* of the question paper X757/76/02 and to the Relationships Sheet X757/76/11.

Care should be taken to give an appropriate number of significant figures in the final answers to calculations.

Write your answers clearly in the spaces provided in this booklet. Additional space for answers and rough work is provided at the end of this booklet. If you use this space you must clearly identify the question number you are attempting. Any rough work must be written in this booklet. You should score through your rough work when you have written your final copy.

Use **blue** or **black** ink.

Before leaving the examination room you must give this booklet to the Invigilator; if you do not, you may lose all the marks for this paper.



The questions for Section 1 are contained in the question paper X757/76/02.

Read these and record your answers on the answer grid on *page 03* opposite.

Use **blue** or **black** ink. Do NOT use gel pens or pencil.

1. The answer to each question is **either** A, B, C, D or E. Decide what your answer is, then fill in the appropriate bubble (see sample question below).
2. There is **only one correct** answer to each question.
3. Any rough working should be done on the additional space for answers and rough work at the end of this booklet.

Sample question

The energy unit measured by the electricity meter in your home is the

- A ampere
- B kilowatt-hour
- C watt
- D coulomb
- E volt.

The correct answer is **B** — kilowatt-hour. The answer **B** bubble has been clearly filled in (see below).

A	B	C	D	E
<input type="radio"/>	<input checked="" type="radio"/>	<input type="radio"/>	<input type="radio"/>	<input type="radio"/>

Changing an answer

If you decide to change your answer, cancel your first answer by putting a cross through it (see below) and fill in the answer you want. The answer below has been changed to **D**.

A	B	C	D	E
<input type="radio"/>	<input checked="" type="radio"/>	<input type="radio"/>	<input checked="" type="radio"/>	<input type="radio"/>

If you then decide to change back to an answer you have already scored out, put a tick (✓) to the **right** of the answer you want, as shown below:

A	B	C	D	E	or	A	B	C	D	E
<input type="radio"/>	<input checked="" type="radio"/>	<input type="radio"/>	<input checked="" type="radio"/>	<input type="radio"/>		<input type="radio"/>	<input checked="" type="radio"/>	<input type="radio"/>	<input type="radio"/>	<input type="radio"/>



SECTION 1 — Answer Grid



	A	B	C	D	E
1	<input type="radio"/>				
2	<input type="radio"/>				
3	<input type="radio"/>				
4	<input type="radio"/>				
5	<input type="radio"/>				
6	<input type="radio"/>				
7	<input type="radio"/>				
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18	<input type="radio"/>				
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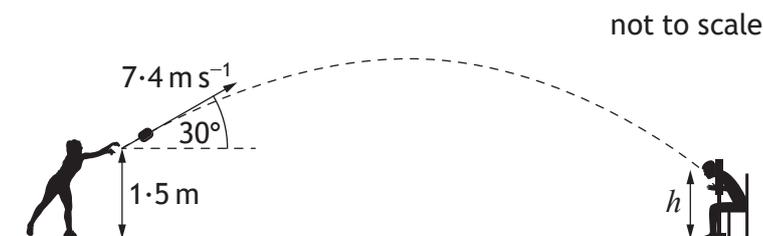
* X 7 5 7 7 6 0 1 0 5 *

SECTION 2 — 110 marks

Attempt ALL questions

1. During a school funfair, a student throws a wet sponge at a teacher. The sponge is thrown with an initial velocity of 7.4 m s^{-1} at an angle of 30° to the horizontal.

The sponge leaves the student's hand at a height of 1.5 m above the ground.



The sponge hits the teacher.

The effects of air resistance can be ignored.

- (a) (i) Calculate:

- (A) the horizontal component of the initial velocity of the sponge; 1
Space for working and answer

- (B) the vertical component of the initial velocity of the sponge. 1
Space for working and answer



1. (a) (continued)

- (ii) Calculate the time taken for the sponge to reach its maximum height.

3

Space for working and answer

- (iii) The sponge takes a further 0.45 s to travel from its maximum height until it hits the teacher.

4

Determine the height h above the ground at which the sponge hits the teacher.

Space for working and answer

- (b) The student throwing the sponge makes the following statement.

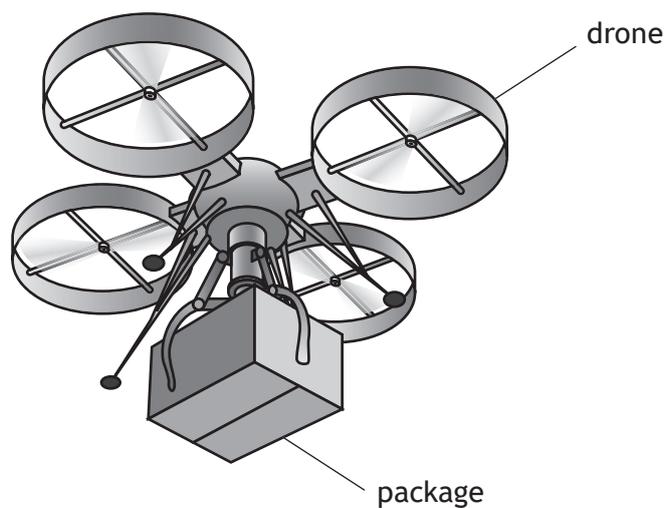
“If the sponge is thrown with a higher speed at the same angle from the same height then it would take a shorter time to hit the teacher in the same place.”

Explain why the student’s statement is incorrect.

2



2. An internet shopping company is planning to use drones to deliver packages.



- (a) During a test the drone is hovering at a constant height above the ground.
The mass of the drone is 5.50 kg.
The mass of the package is 1.25 kg.
- (i) Determine the upward force produced by the drone.
- Space for working and answer*

3

2. (a) (continued)

(ii) The package is now lowered using a motor and a cable.

A battery supplies 12 V across the motor. The resistance of the motor is 9.6Ω .

Calculate the power dissipated by the motor.

3

Space for working and answer

(iii) While the package is being lowered the cable breaks.

The upward force produced by the drone remains constant.

Describe the vertical motion of the drone immediately after the cable breaks.

2

Justify your answer.

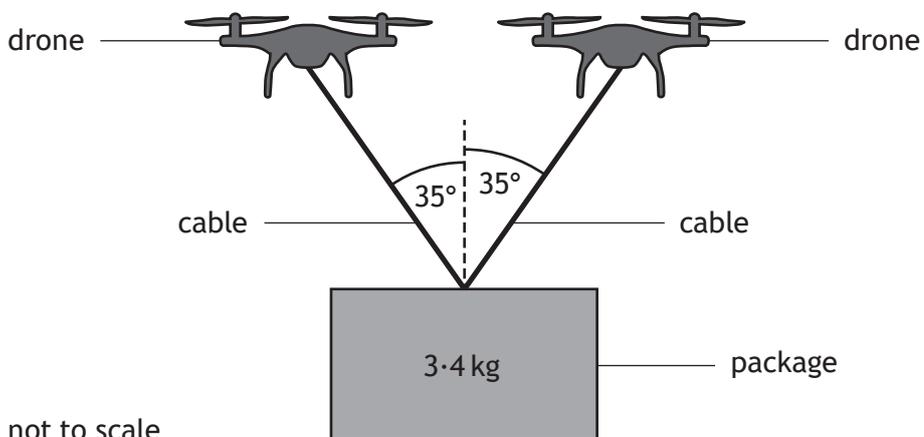
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* X 7 5 7 7 6 0 1 0 9 *

2. (continued)

(b) To carry a package with a greater mass two drones are used as shown.



The drones are hovering at a constant height above the ground.
 The mass of the package suspended from the two drones is 3.4 kg.
 Determine the tension in each cable.

Space for working and answer

4

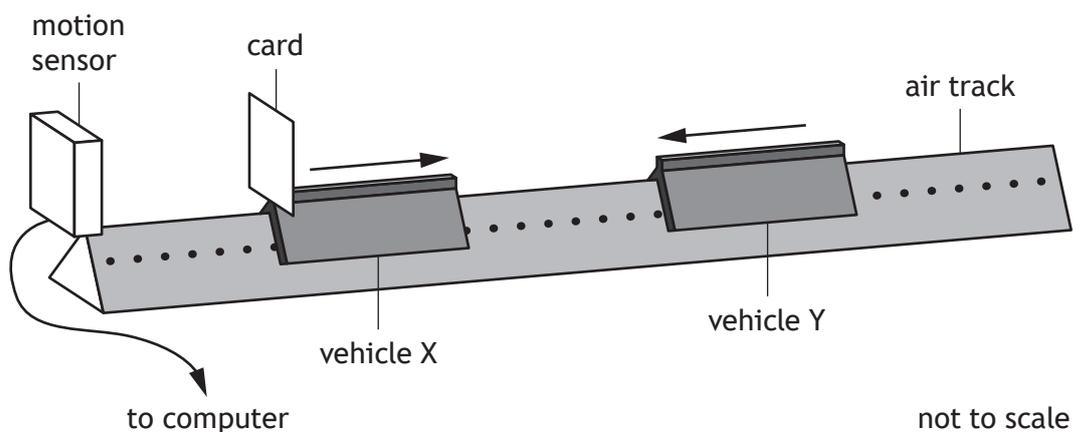
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* X 7 5 7 7 6 0 1 1 1 *

3. A student sets up an experiment to investigate a collision between two vehicles on a frictionless air track.

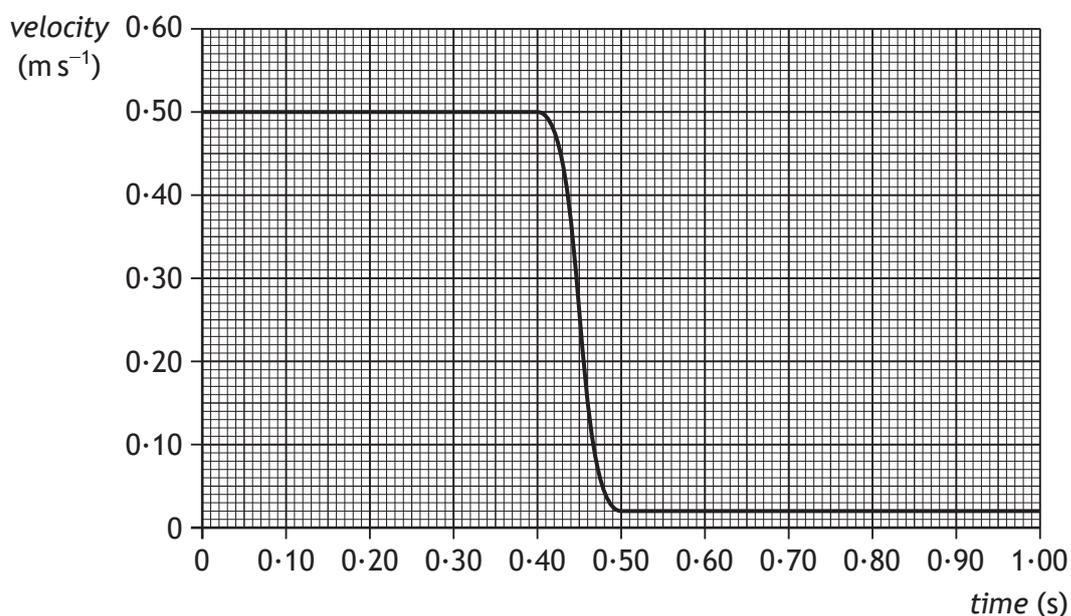


Vehicle X of mass 0.75 kg is travelling to the right along the track.

Vehicle Y of mass 0.50 kg is travelling to the left along the track with a speed of 0.30 m s^{-1} .

The vehicles collide and move off separately.

A computer displays a graph showing the velocity of vehicle X from just before the collision to just after the collision.



MARKS

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3. (continued)

(a) Show that the velocity of vehicle Y after the collision is 0.42 m s^{-1} .

2

Space for working and answer

(b) Determine the impulse on vehicle Y during the collision.

3

Space for working and answer

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* X 7 5 7 7 6 0 1 1 3 *

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3. (continued)

- (c) Explain how the student would determine whether the collision was elastic or inelastic.

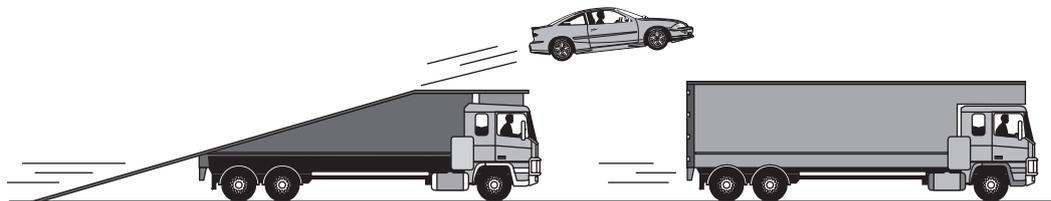
2



* X 7 5 7 7 6 0 1 1 4 *

4. A stunt is being carried out during the making of a film.

A car is to be driven up a ramp on a moving lorry by a stunt driver, who will attempt to land the car safely on the roof of a second moving lorry. The car is to stop on the roof of the second lorry while this lorry is still moving.



Using your knowledge of physics, comment on the challenges involved in carrying out the stunt successfully.

3

[Turn over



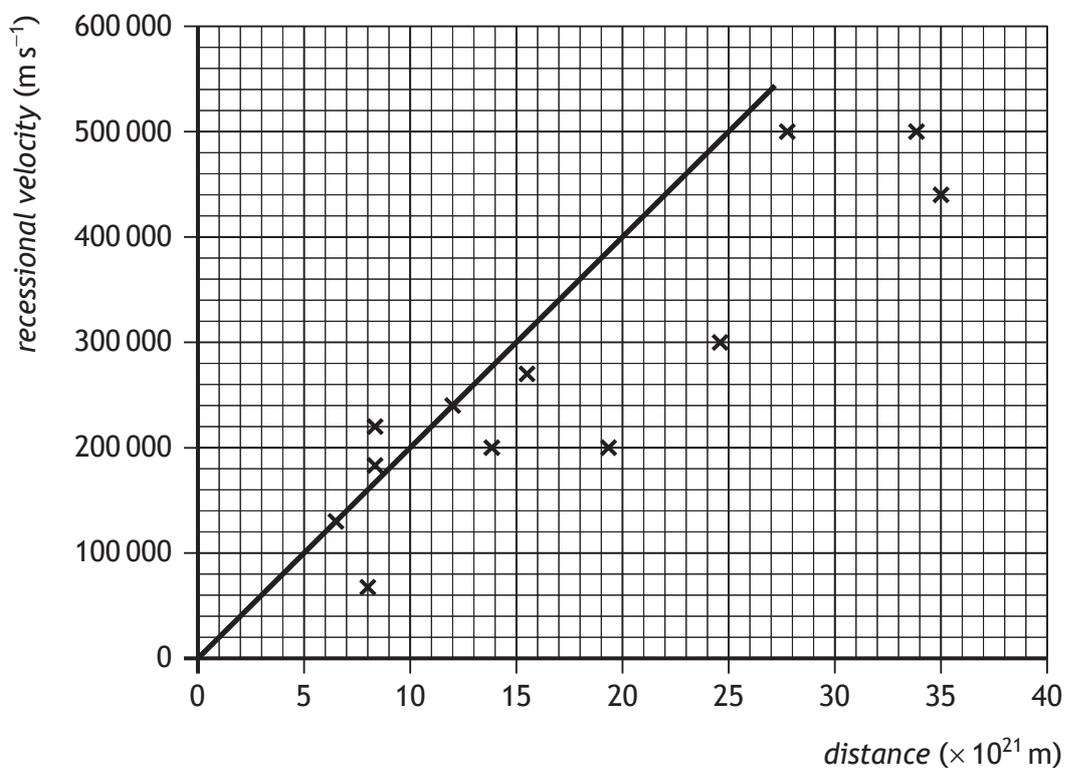
* X 7 5 7 7 6 0 1 1 5 *

5. Hubble's Law states that the universe is expanding. The expanding universe is one piece of evidence that supports the Big Bang theory.

(a) State one other piece of evidence that supports the Big Bang theory.

1

(b) A student plots some of the original data from the 1929 paper by Edwin Hubble and adds the line shown in order to determine a value for the Hubble constant H_0 .



The student calculates the gradient of their line and obtains a value for the Hubble constant of $2.0 \times 10^{-17} \text{ s}^{-1}$.

The age of the universe can be calculated using the relationship

$$\text{age of universe} = \frac{1}{H_0}$$



5. (b) (continued)

- (i) Calculate the age of the universe, in years, obtained when using the student's value for the Hubble constant.

2

Space for working and answer

- (ii) The current estimate for the age of the universe is 13.8×10^9 years.

- (A) State why the value obtained in (b)(i) is different from the current estimate for the age of the universe.

1

- (B) Suggest a change that the student could make to their graph to obtain a value closer to the current estimate for the age of the universe.

1

- (c) It has been discovered that the rate of expansion of the universe is increasing.

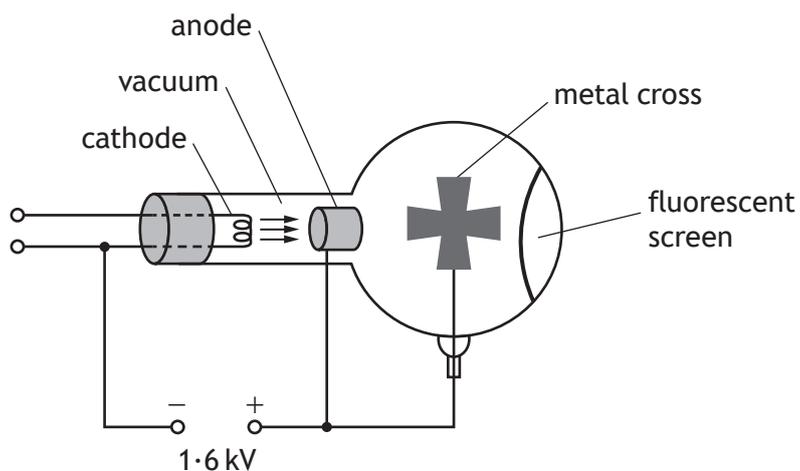
State what physicists think is responsible for this increase.

1



* X 7 5 7 7 6 0 1 1 7 *

6. An experiment is set up to demonstrate a simple particle accelerator.



(a) Electrons are accelerated from rest between the cathode and the anode by a potential difference of 1.6 kV.

(i) Show that the work done in accelerating an electron from rest is $2.6 \times 10^{-16} \text{ J}$.

2

Space for working and answer

(ii) Calculate the speed of the electron as it reaches the anode.

3

Space for working and answer



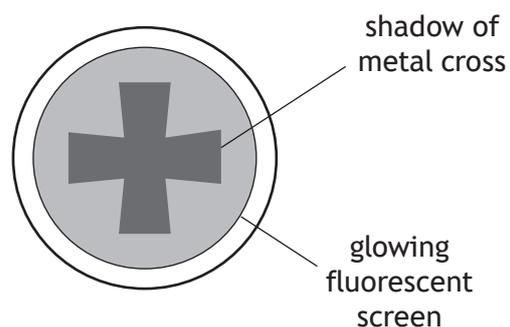
6. (continued)

- (b) As the electrons travel through the vacuum towards the fluorescent screen they spread out.

In the path of the electrons there is a metal cross, which is connected to the positive terminal of the supply. The electrons that hit the cross are stopped by the metal.

Electrons that get past the metal cross hit a fluorescent screen at the far side of the tube.

When electrons hit the fluorescent screen, the screen glows.



The potential difference between the anode and the cathode is now increased to 2.2 kV. This changes what is observed on the screen.

Suggest one change that is observed.

2

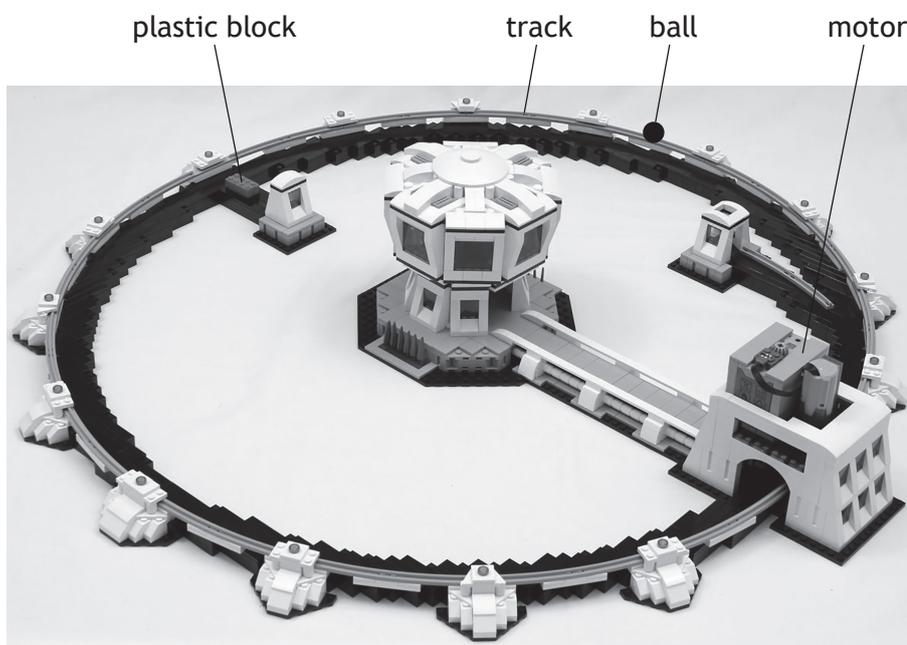
You must justify your answer.

[Turn over



6. (continued)

- (c) A student builds a model of a particle accelerator. The model accelerates a small ball on a circular track. A battery-operated motor accelerates the ball each time it passes the motor. To cause a collision a plastic block is pushed onto the track. The ball then hits the block.



Using your knowledge of physics comment on the model compared to a real particle accelerator, such as the large hadron collider at CERN.

3



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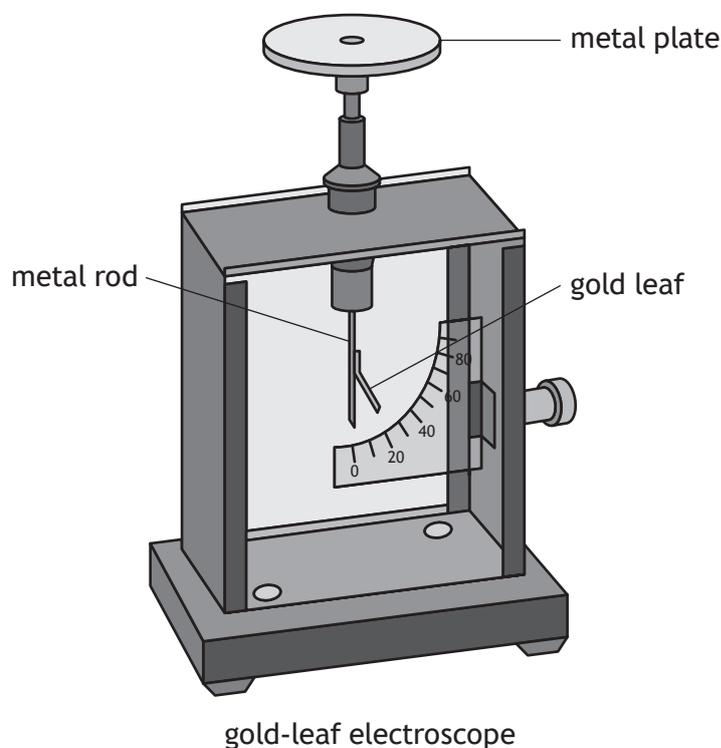
6. (c) (continued)



* X 7 5 7 7 6 0 1 2 1 *

7. A student uses a gold-leaf electroscope to investigate the photoelectric effect. A deflection of the gold leaf on the electroscope shows that the metal plate is charged.

The student charges the metal plate on the electroscope and the gold leaf is deflected.



- (a) Ultraviolet light is shone onto the negatively charged metal plate. The gold-leaf electroscope does not discharge. This indicates that photoelectrons are not ejected from the surface of the metal.

Suggest one reason why photoelectrons are not ejected from the surface of the metal.

1

7. (continued)

- (b) The student adjusts the experiment so that the gold-leaf electroscope now discharges when ultraviolet light is shone onto the plate.

The work function for the metal plate is 6.94×10^{-19} J.

- (i) State what is meant by a *work function of 6.94×10^{-19} J*.

1

- (ii) The irradiance of the ultraviolet light on the metal plate is reduced by increasing the distance between the gold-leaf electroscope and the ultraviolet light source.

State what effect, if any, this has on the maximum kinetic energy of the photoelectrons ejected from the surface of the metal.

2

Justify your answer.

[Turn over

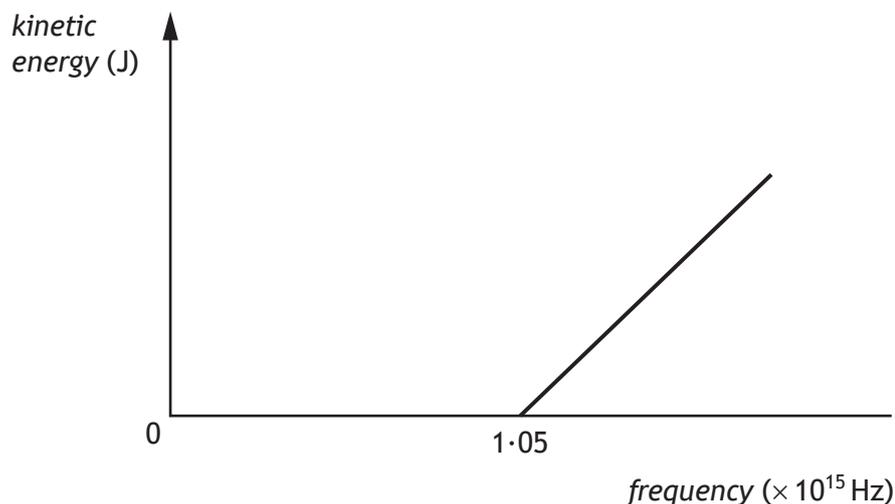


* X 7 5 7 7 6 0 1 2 3 *

7. (continued)

- (c) The graph shows how the kinetic energy of the photoelectrons ejected from the metal plate varies as the frequency of the incident radiation increases.

The threshold frequency for the metal plate is 1.05×10^{15} Hz.



The metal plate is now replaced with a different metal plate made of aluminium.

The aluminium has a threshold frequency of 0.99×10^{15} Hz.

Add a line to the graph to show how the kinetic energy of the photoelectrons ejected from the aluminium plate varies as the frequency of the incident radiation increases.

2

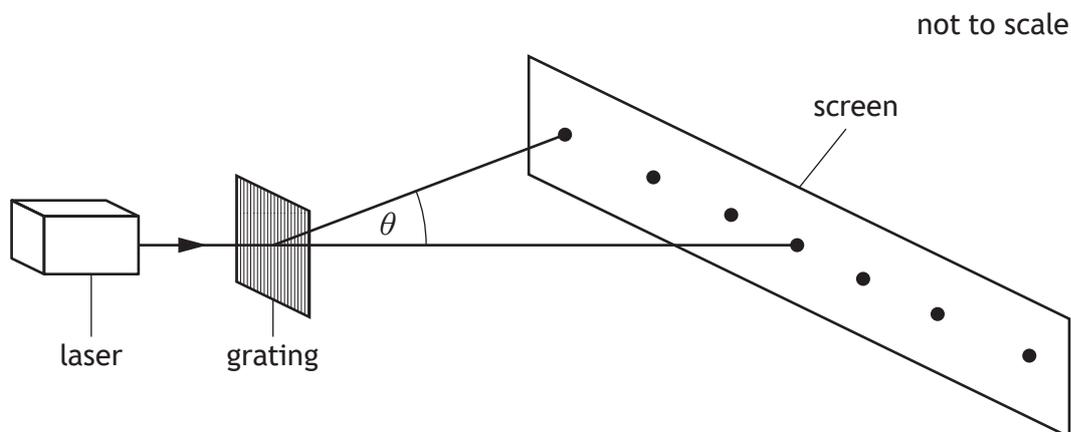
(An additional graph, if required, can be found on *page 45*.)

- (d) Explain why the photoelectric effect provides evidence for the particle nature of light.

1



8. A student investigates interference of light by directing laser light of wavelength 630 nm onto a grating as shown.



- (a) A pattern of bright spots is observed on a screen.
- (i) Explain, in terms of waves, how bright spots are produced on the screen. 1

- (ii) The grating has 250 lines per millimetre.
- Calculate the angle θ between the central maximum and the third order maximum. 3
- Space for working and answer*

[Turn over



8. (a) (continued)

- (iii) The grating is now replaced by one which has 600 lines per millimetre.

State the effect of this change on the pattern observed.

2

Justify your answer.

- (iv) The interference pattern is produced by coherent light.

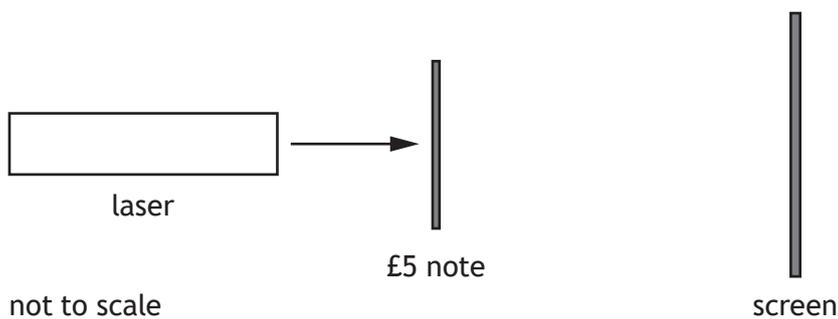
State what is meant by the term *coherent*.

1



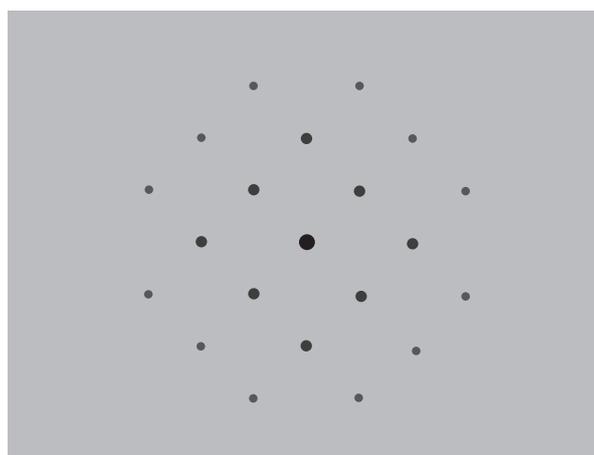
8. (continued)

(b) The student now shines light from the laser onto a £5 note.



When it is shone through the transparent section of the note the student observes a pattern of bright spots on the screen.

The diagram below shows the pattern that the student observes on the screen.

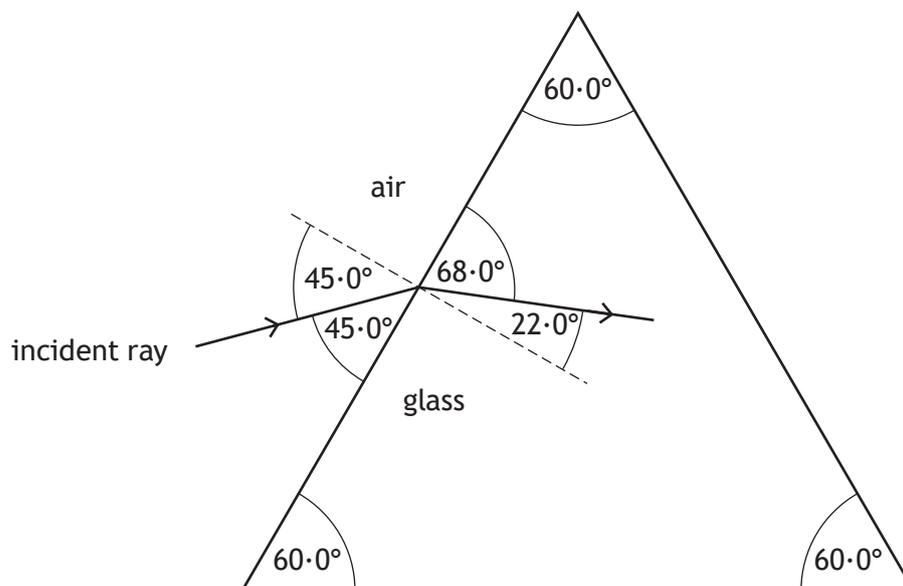


Suggest a reason for the difference in the pattern produced using the £5 note and the pattern produced using the grating.

1

[Turn over

9. A ray of monochromatic light is incident on a glass prism as shown.



(a) Show that the refractive index of the glass for this ray of light is 1.89. 2

Space for working and answer

(b) (i) State what is meant by the term *critical angle*. 1



9. (b) (continued)

(ii) Calculate the critical angle for this light in the prism.

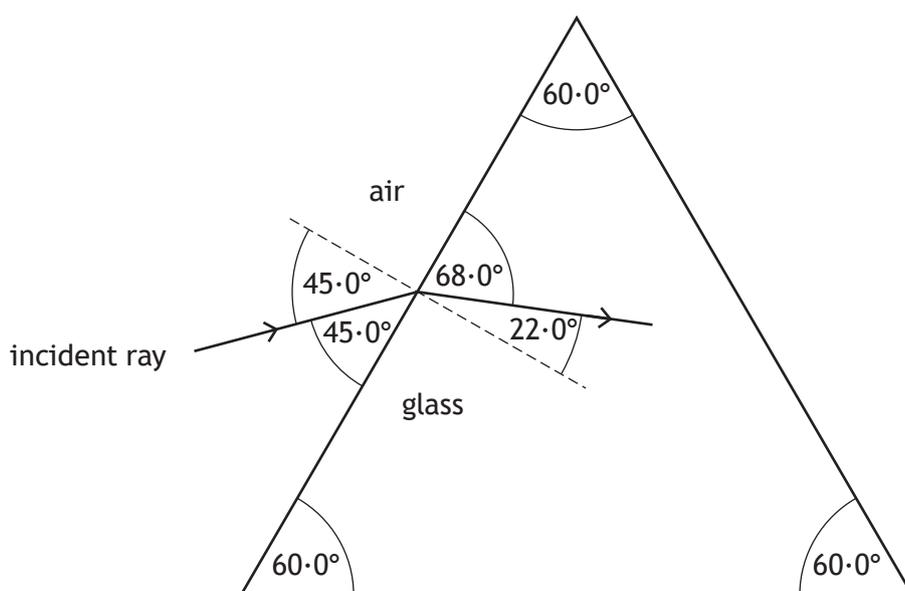
3

Space for working and answer

(iii) Complete the diagram below to show the path of the ray as it passes through the prism and emerges into the air.

4

Mark on the diagram the values of all relevant angles.



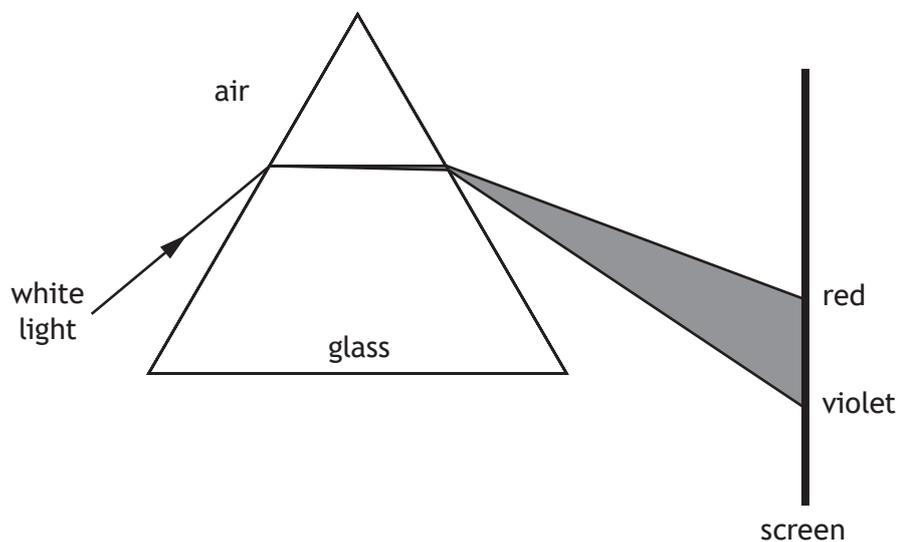
(An additional diagram, if required, can be found on page 45.)

[Turn over



9. (continued)

(c) A ray of white light is shone through the prism and a spectrum is observed as shown.



The prism is now replaced with another prism made from a different type of glass with a lower refractive index.

Describe one difference in the spectrum produced by this prism compared to the spectrum produced by the first prism.

1

10. In a laboratory experiment, light from a hydrogen discharge lamp is used to produce a line emission spectrum. The line spectrum for hydrogen has four lines in the visible region as shown.



- (a) The production of the line spectrum can be explained using the Bohr model of the atom.

State **two** features of the *Bohr model* of the atom.

2

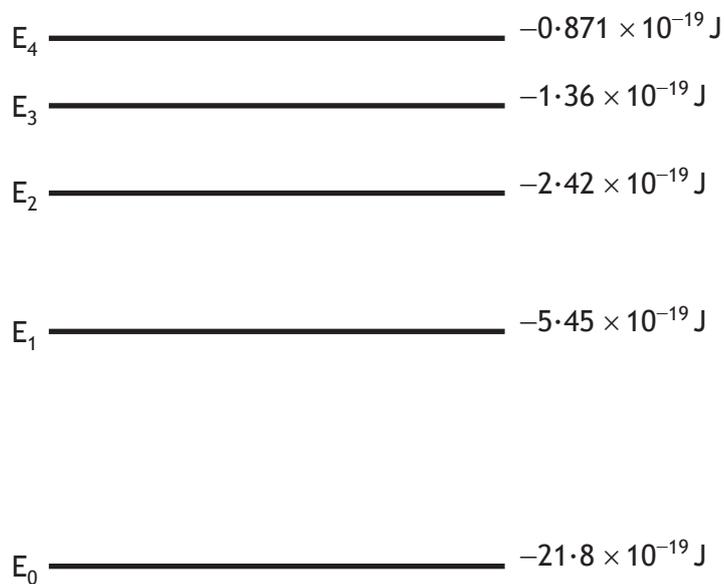
[Turn over



* X 7 5 7 7 6 0 1 3 1 *

10. (continued)

(b) Some of the energy levels of the hydrogen atom are shown.



One of the spectral lines is due to electron transitions from E_3 to E_1 .

Determine the frequency of the photon emitted when an electron makes this transition.

3

Space for working and answer



10. (continued)

- (c) In the laboratory, a line in the hydrogen spectrum is observed at a wavelength of 656 nm.

When the spectrum of light from a distant galaxy is viewed, this hydrogen line is now observed at a wavelength of 661 nm.

Determine the recessional velocity of the distant galaxy.

5

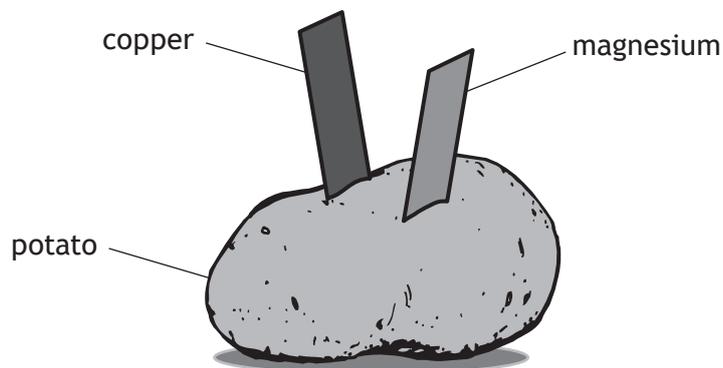
Space for working and answer

[Turn over

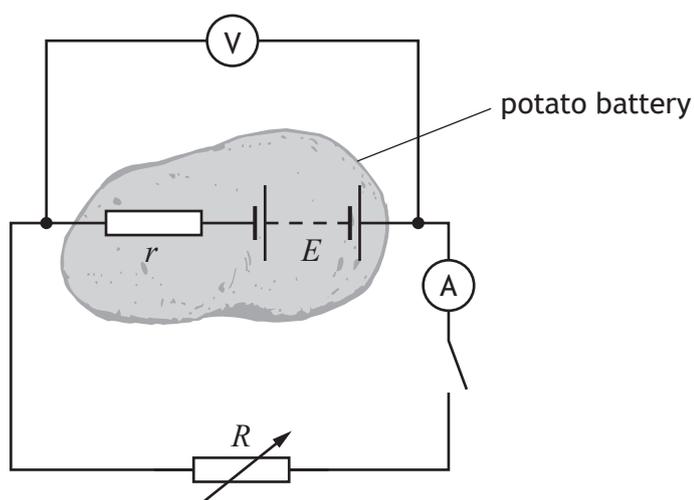


* X 7 5 7 7 6 0 1 3 3 *

11. A student constructs a battery using a potato, a strip of copper and a strip of magnesium.



The student then sets up the following circuit with the potato battery connected to a variable resistor R , in order that the electromotive force (e.m.f.) and internal resistance of the battery may be determined.



- (a) State what is meant by the term *electromotive force* (e.m.f.).

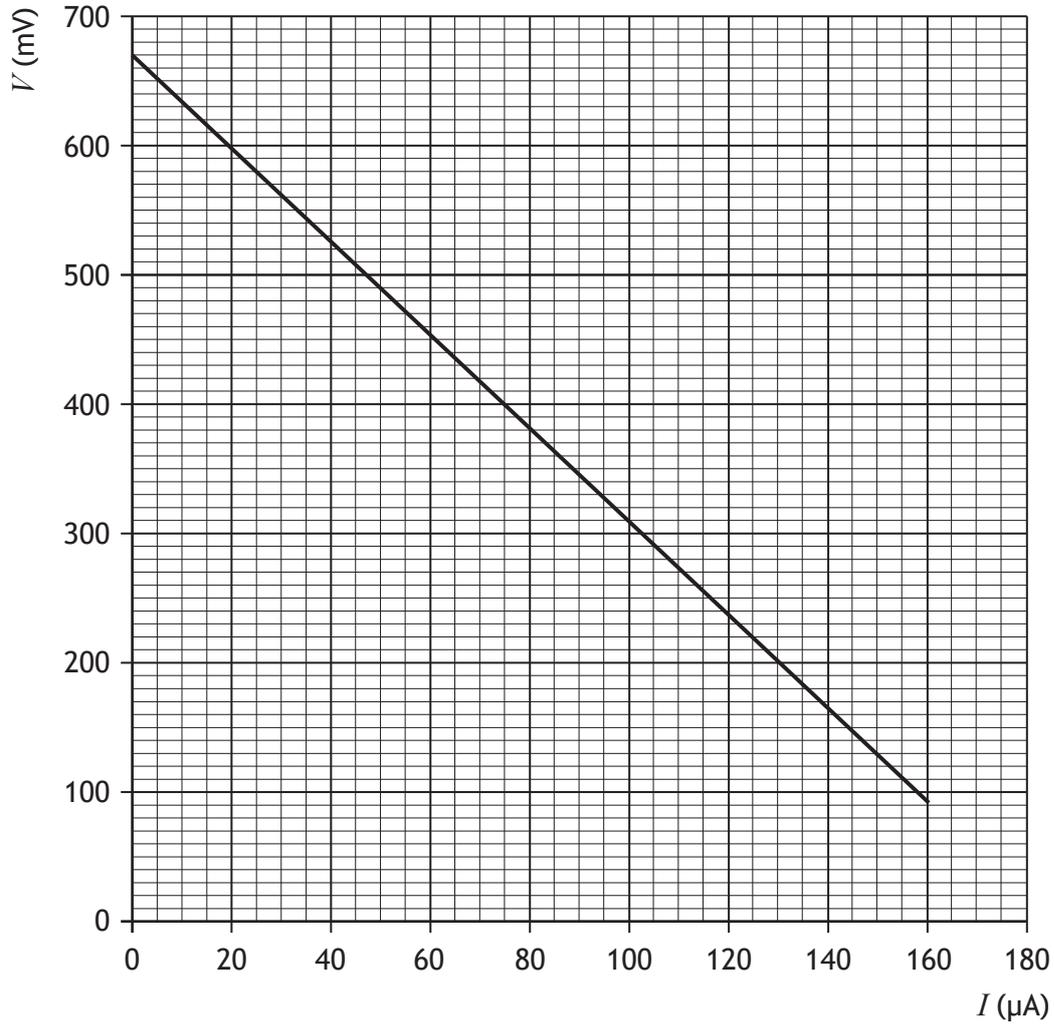
1

11. (continued)

MARKS

DO NOT
WRITE IN
THIS
MARGIN

- (b) The student uses readings of current I and terminal potential difference V from this circuit to produce the graph shown.



Determine the internal resistance of the potato battery.

3

Space for working and answer

[Turn over



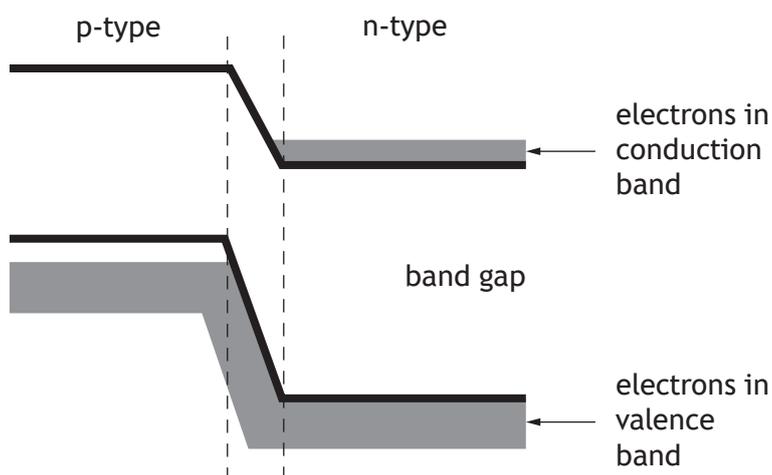
* X 7 5 7 7 6 0 1 3 5 *

11. (continued)

- (c) The student connects a red LED and a blue LED, in turn, to the battery. The LEDs are forward biased when connected.

The student observes that the battery will operate the red LED but not the blue LED.

The diagram represents the band structure of the blue LED.



LEDs emit light when electrons fall from the conduction band into the valence band of the p-type semiconductor.

Explain, using **band theory**, why the blue LED will not operate with this battery.

1

[Turn over for next question

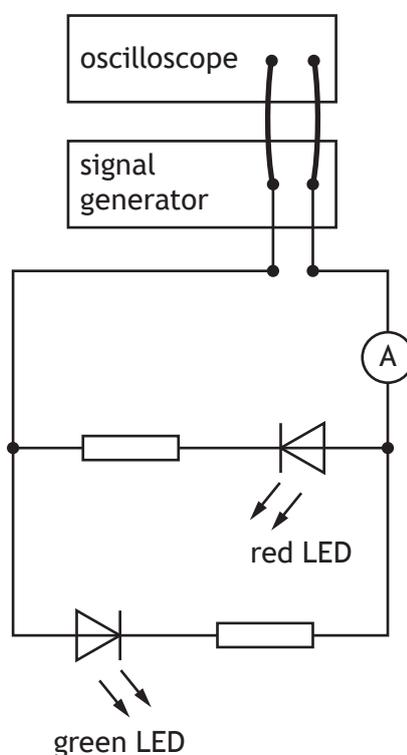
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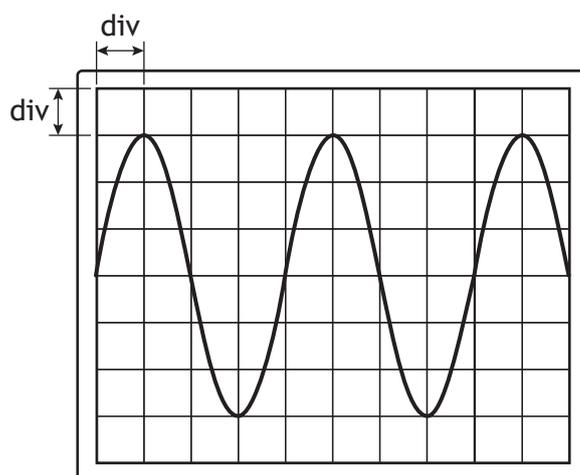
* X 7 5 7 7 6 0 1 3 7 *

12. A student carries out a series of experiments to investigate alternating current.

(a) A signal generator is connected to an oscilloscope and a circuit as shown.



The output of the signal generator is displayed on the oscilloscope.



The Y-gain setting on the oscilloscope is 1.0 V/div .

The timebase setting on the oscilloscope is 0.5 s/div .



12. (a) (continued)

(i) Determine the peak voltage of the output of the signal generator. 1
Space for working and answer

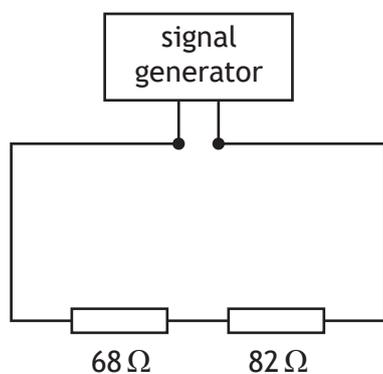
(ii) Determine the frequency of the output of the signal generator. 3
Space for working and answer

(iii) The student observes that the red LED is only lit when the ammeter gives a positive reading and the green LED is only lit when the ammeter gives a negative reading. 2
 Explain these observations.



12. (continued)

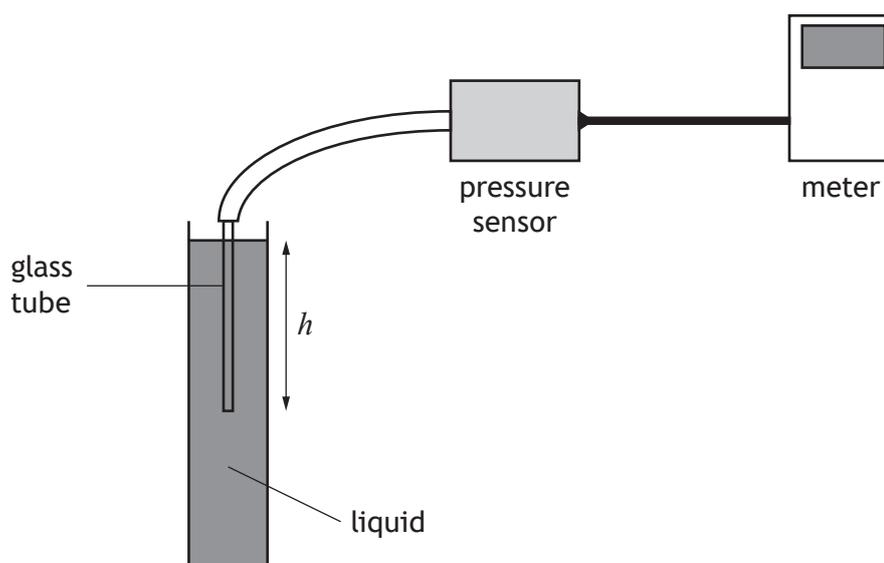
- (b) The signal generator is now connected in a circuit as shown.
 The settings on the signal generator are unchanged.
 The signal generator has negligible internal resistance.



Determine the r.m.s. voltage across the 82 Ω resistor.
 Space for working and answer

5

13. A student sets up an experiment to investigate the pressure due to a liquid as shown.



The pressure due to a liquid is given by the relationship

$$p = \rho gh$$

where p is the pressure due to the liquid in pascals (Pa),

g is the gravitational field strength in N kg^{-1} ,

ρ is the density of the liquid in kg m^{-3} ,

and h is the depth in the liquid in m.

- (a) The student initially carries out the investigation using water.

The density of water is $1.00 \times 10^3 \text{ kg m}^{-3}$.

Calculate the pressure due to the water at a depth of 0.35 m.

2

Space for working and answer



13. (continued)

- (b) The student repeats the experiment with a different liquid.

The pressure meter is set to zero before the glass tube is lowered into the liquid.

The student takes measurements of the pressure at various depths below the surface of the liquid.

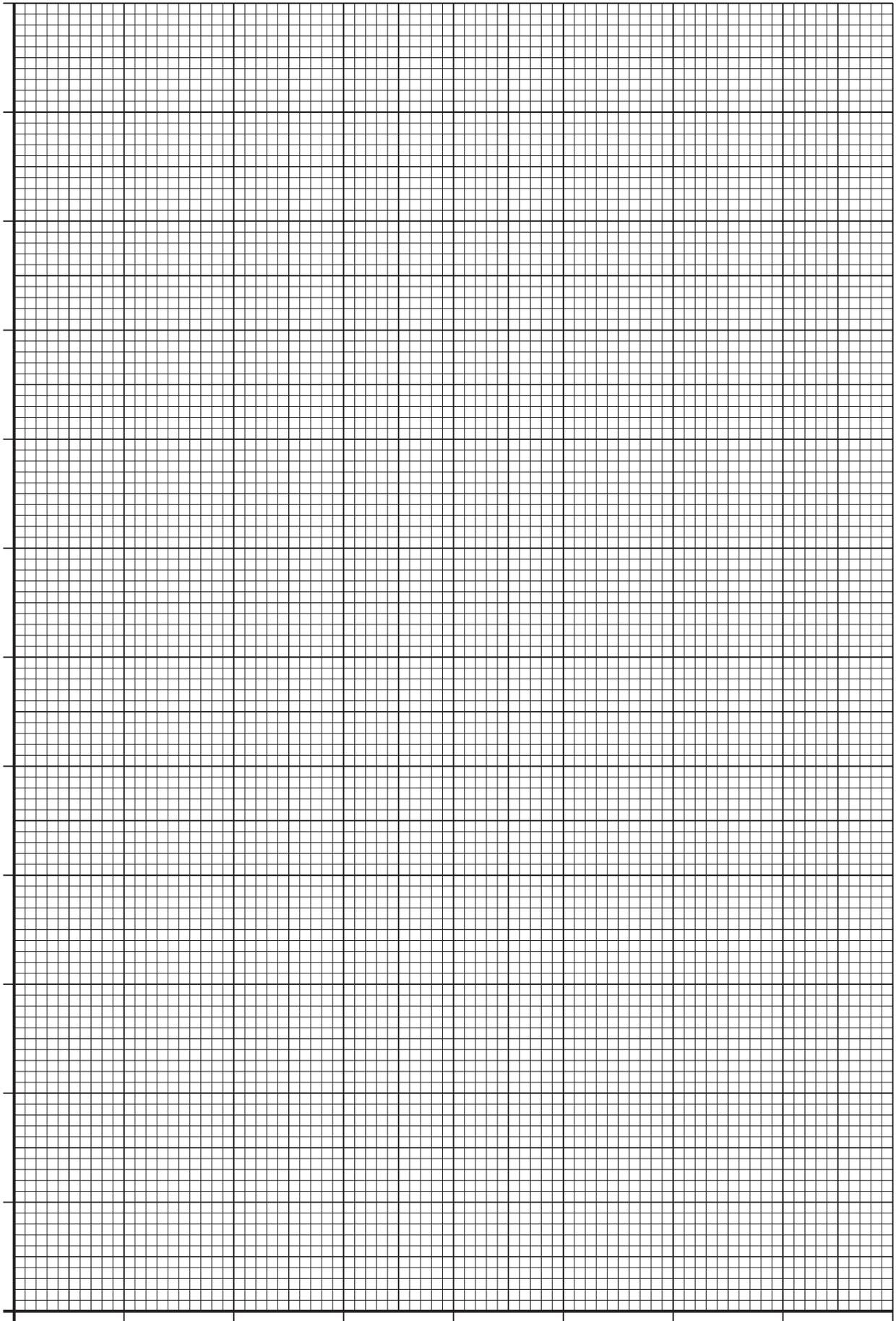
The student records the following information.

Depth, h (m)	Pressure, p (kPa)
0.10	1.2
0.20	2.5
0.30	3.6
0.40	4.9
0.50	6.2

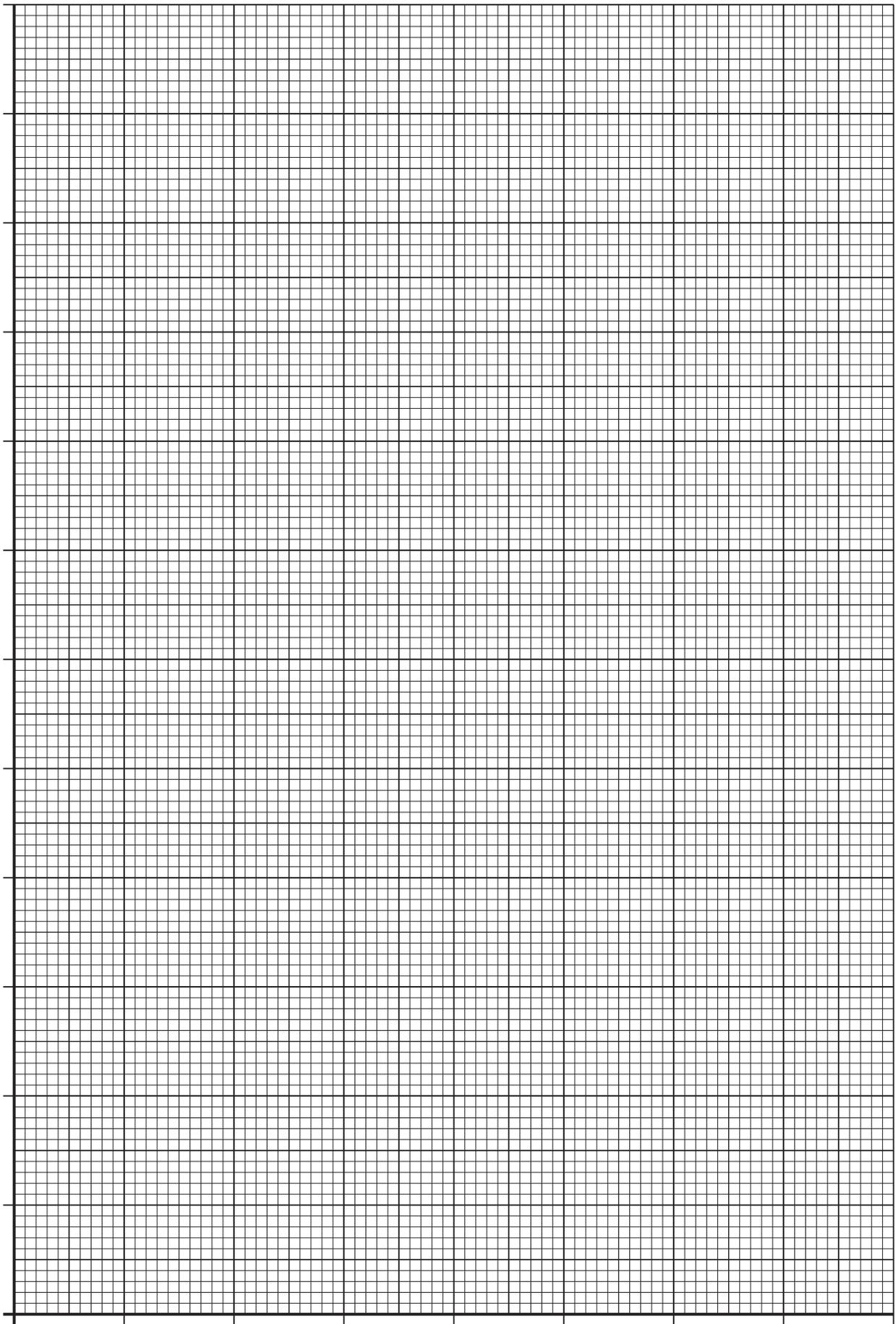
- (i) Using the square-ruled paper on *page 43*, draw a graph of p against h . 3
(Additional graph paper, if required, can be found on *page 44*.)
- (ii) Calculate the gradient of your graph. 2
Space for working and answer
- (iii) Determine the density of this liquid. 2
Space for working and answer

[END OF QUESTION PAPER]





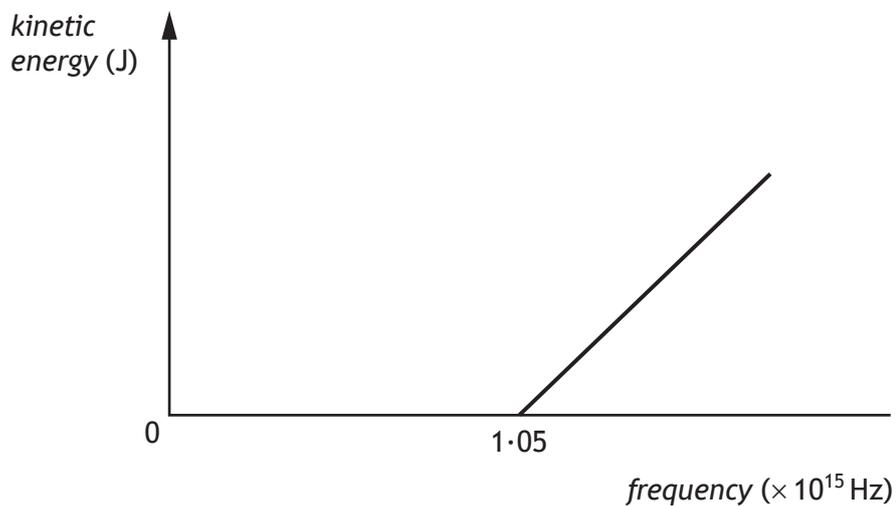
* X 7 5 7 7 6 0 1 4 3 *



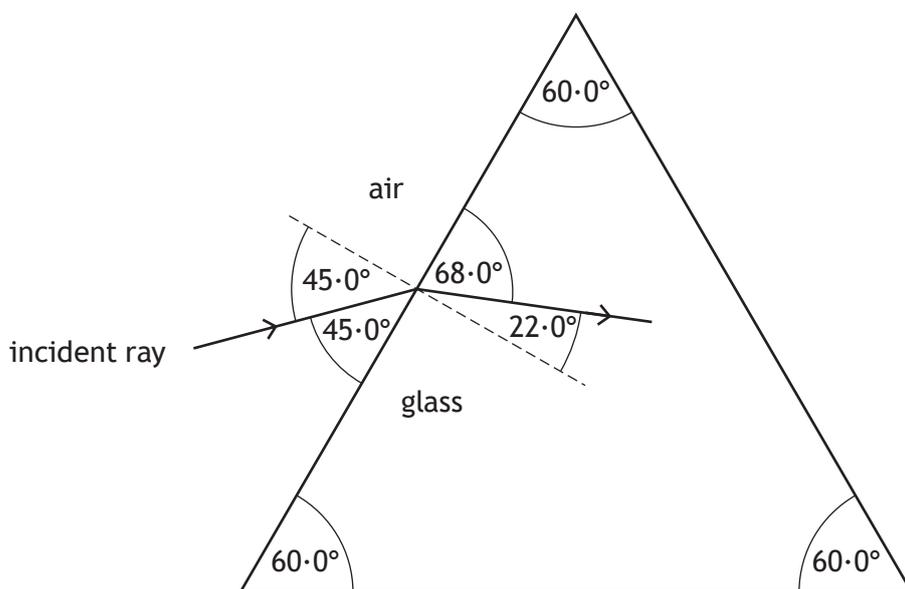
* X 7 5 7 7 6 0 1 4 4 *

ADDITIONAL SPACE FOR ANSWERS AND ROUGH WORK

Additional graph for use with Question 7 (c)



Additional diagram for use with Question 9 (b) (iii)



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ADDITIONAL SPACE FOR ANSWERS AND ROUGH WORK



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ACKNOWLEDGEMENTS

Question 1 – Snap2Art/Shutterstock.com

Question 2 (b) – Studio Caramel/Shutterstock.com

Question 6 (c) – Image is taken from <http://jkbrickworks.com/jkbw/wp-content/uploads/2014/11/accelerator.jpg?x84406>. Reproduced by kind permission of Jason Allemann.



* X 7 5 7 7 6 0 1 4 8 *



National
Qualifications
2018

X757/76/11

**Physics
Relationships Sheet**

TUESDAY, 8 MAY

9:00 AM – 11:30 AM



* X 7 5 7 7 6 1 1 *

Relationships required for Physics Higher

$$d = \bar{v}t$$

$$s = \bar{v}t$$

$$v = u + at$$

$$s = ut + \frac{1}{2}at^2$$

$$v^2 = u^2 + 2as$$

$$s = \frac{1}{2}(u + v)t$$

$$W = mg$$

$$F = ma$$

$$E_w = Fd$$

$$E_p = mgh$$

$$E_k = \frac{1}{2}mv^2$$

$$P = \frac{E}{t}$$

$$p = mv$$

$$Ft = mv - mu$$

$$F = G \frac{m_1 m_2}{r^2}$$

$$t' = \frac{t}{\sqrt{1 - (v/c)^2}}$$

$$l' = l \sqrt{1 - (v/c)^2}$$

$$f_o = f_s \left(\frac{v}{v \pm v_s} \right)$$

$$z = \frac{\lambda_{\text{observed}} - \lambda_{\text{rest}}}{\lambda_{\text{rest}}}$$

$$z = \frac{v}{c}$$

$$v = H_0 d$$

$$W = QV$$

$$E = mc^2$$

$$E = hf$$

$$E_k = hf - hf_0$$

$$E_2 - E_1 = hf$$

$$T = \frac{1}{f}$$

$$v = f\lambda$$

$$d \sin \theta = m\lambda$$

$$n = \frac{\sin \theta_1}{\sin \theta_2}$$

$$\frac{\sin \theta_1}{\sin \theta_2} = \frac{\lambda_1}{\lambda_2} = \frac{v_1}{v_2}$$

$$\sin \theta_c = \frac{1}{n}$$

$$I = \frac{k}{d^2}$$

$$I = \frac{P}{A}$$

$$V_{\text{peak}} = \sqrt{2}V_{\text{rms}}$$

$$I_{\text{peak}} = \sqrt{2}I_{\text{rms}}$$

$$Q = It$$

$$V = IR$$

$$P = IV = I^2 R = \frac{V^2}{R}$$

$$R_T = R_1 + R_2 + \dots$$

$$\frac{1}{R_T} = \frac{1}{R_1} + \frac{1}{R_2} + \dots$$

$$E = V + Ir$$

$$V_1 = \left(\frac{R_1}{R_1 + R_2} \right) V_s$$

$$\frac{V_1}{V_2} = \frac{R_1}{R_2}$$

$$C = \frac{Q}{V}$$

$$E = \frac{1}{2}QV = \frac{1}{2}CV^2 = \frac{1}{2} \frac{Q^2}{C}$$

$$\text{path difference} = m\lambda \quad \text{or} \quad \left(m + \frac{1}{2}\right)\lambda \quad \text{where } m = 0, 1, 2, \dots$$

$$\text{random uncertainty} = \frac{\text{max. value} - \text{min. value}}{\text{number of values}}$$

Additional Relationships

Circle

$$\text{circumference} = 2\pi r$$

$$\text{area} = \pi r^2$$

Sphere

$$\text{area} = 4\pi r^2$$

$$\text{volume} = \frac{4}{3}\pi r^3$$

Trigonometry

$$\sin \theta = \frac{\text{opposite}}{\text{hypotenuse}}$$

$$\cos \theta = \frac{\text{adjacent}}{\text{hypotenuse}}$$

$$\tan \theta = \frac{\text{opposite}}{\text{adjacent}}$$

$$\sin^2 \theta + \cos^2 \theta = 1$$

Electron Arrangements of Elements

Group 1 Group 2
(1)

1 H	4 Be
Hydrogen 1	(2)
3 Li	2,2 B
2,1 Lithium	Beryllium
11 Na	12 Mg
2,8,1 Sodium	2,8,2 Magnesium
19 K	20 Ca
2,8,8,1 Potassium	2,8,8,2 Calcium
37 Rb	38 Sr
2,8,18,8,1 Rubidium	2,8,18,8,2 Strontium
55 Cs	56 Ba
2,8,18,18,8,1 Caesium	2,8,18,18,8,2 Barium
87 Fr	88 Ra
2,8,18,32,18,8,1 Francium	2,8,18,32,18,8,2 Radium

Key

Atomic number
Symbol
Electron arrangement
Name

Transition Elements

21 Sc	22 Ti	23 V	24 Cr	25 Mn	26 Fe	27 Co	28 Ni	29 Cu	30 Zn
Scandium	Titanium	Vanadium	Chromium	Manganese	Iron	Cobalt	Nickel	Copper	Zinc
39 Y	40 Zr	41 Nb	42 Mo	43 Tc	44 Ru	45 Rh	46 Pd	47 Ag	48 Cd
2,8,18,9,2 Yttrium	2,8,18,10,2 Zirconium	2,8,18,12,1 Niobium	2,8,18,13,1 Molybdenum	2,8,18,13,2 Technetium	2,8,18,15,1 Ruthenium	2,8,18,16,1 Rhodium	2,8,18,18,0 Palladium	2,8,18,18,1 Silver	2,8,18,18,2 Cadmium
57 La	72 Hf	73 Ta	74 W	75 Re	76 Os	77 Ir	78 Pt	79 Au	80 Hg
2,8,18,18,9,2 Lanthanum	2,8,18,32,10,2 Hafnium	2,8,18,32,11,2 Tantalum	2,8,18,32,12,2 Tungsten	2,8,18,32,13,2 Rhenium	2,8,18,32,14,2 Osmium	2,8,18,32,15,2 Iridium	2,8,18,32,17,1 Platinum	2,8,18,32,18,1 Gold	2,8,18,32,18,2 Mercury
89 Ac	104 Rf	105 Db	106 Sg	107 Bh	108 Hs	109 Mt	110 Ds	111 Rg	112 Cn
2,8,18,32,18,9,2 Actinium	2,8,18,32,32,10,2 Rutherfordium	2,8,18,32,32,11,2 Dubnium	2,8,18,32,32,12,2 Seaborgium	2,8,18,32,32,13,2 Bohrium	2,8,18,32,32,14,2 Hassium	2,8,18,32,32,15,2 Meitnerium	2,8,18,32,32,17,1 Darmstadtium	2,8,18,32,32,18,1 Roentgenium	2,8,18,32,32,18,2 Copernicium

Group 3 Group 4 Group 5 Group 6 Group 7 Group 8
(18)

5 B	6 C	7 N	8 O	9 F	10 Ne
2,3 Boron	2,4 Carbon	2,5 Nitrogen	2,6 Oxygen	2,7 Fluorine	2,8 Neon
13 Al	14 Si	15 P	16 S	17 Cl	18 Ar
2,8,3 Aluminium	2,8,4 Silicon	2,8,5 Phosphorus	2,8,6 Sulfur	2,8,7 Chlorine	2,8,8 Argon
31 Ga	32 Ge	33 As	34 Se	35 Br	36 Kr
2,8,18,3 Gallium	2,8,18,4 Germanium	2,8,18,5 Arsenic	2,8,18,6 Selenium	2,8,18,7 Bromine	2,8,18,8 Krypton
49 In	50 Sn	51 Sb	52 Te	53 I	54 Xe
2,8,18,18,3 Indium	2,8,18,18,4 Tin	2,8,18,18,5 Antimony	2,8,18,18,6 Tellurium	2,8,18,18,7 Iodine	2,8,18,18,8 Xenon
81 Tl	82 Pb	83 Bi	84 Po	85 At	86 Rn
2,8,18,32,18,3 Thallium	2,8,18,32,18,4 Lead	2,8,18,32,18,5 Bismuth	2,8,18,32,18,6 Polonium	2,8,18,32,18,7 Astatine	2,8,18,32,18,8 Radon

Lanthanides

57 La	58 Ce	59 Pr	60 Nd	61 Pm	62 Sm	63 Eu	64 Gd	65 Tb	66 Dy	67 Ho	68 Er	69 Tm	70 Yb	71 Lu
2,8,18,18,9,2 Lanthanum	2,8,18,20,8,2 Cerium	2,8,18,21,8,2 Praseodymium	2,8,18,22,8,2 Neodymium	2,8,18,23,8,2 Promethium	2,8,18,24,8,2 Samarium	2,8,18,25,8,2 Europium	2,8,18,25,9,2 Gadolinium	2,8,18,27,8,2 Terbium	2,8,18,28,8,2 Dysprosium	2,8,18,29,8,2 Holmium	2,8,18,30,8,2 Erbium	2,8,18,31,8,2 Thulium	2,8,18,32,8,2 Ytterbium	2,8,18,32,9,2 Lutetium

Actinides

89 Ac	90 Th	91 Pa	92 U	93 Np	94 Pu	95 Am	96 Cm	97 Bk	98 Cf	99 Es	100 Fm	101 Md	102 No	103 Lr
2,8,18,32,18,9,2 Actinium	2,8,18,32,18,10,2 Thorium	2,8,18,32,20,9,2 Protactinium	2,8,18,32,21,9,2 Uranium	2,8,18,32,22,9,2 Neptunium	2,8,18,32,24,8,2 Plutonium	2,8,18,32,25,8,2 Americium	2,8,18,32,25,9,2 Curium	2,8,18,32,27,8,2 Berkelium	2,8,18,32,28,8,2 Californium	2,8,18,32,29,8,2 Einsteinium	2,8,18,32,30,8,2 Fermium	2,8,18,32,31,8,2 Mendelevium	2,8,18,32,32,8,2 Nobelium	2,8,18,32,32,9,2 Lawrencium